Characterization of starter-free Queso Fresco made with sodium-potassium salt blends over 12 weeks of 4°C storage

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ABSTRACT

Development of reduced-sodium cheese to meet the demands of consumers concerned about sodium levels in their diet is challenging when a high-moisture, higher pH, fresh cheese, such as Queso Fresco (QF), depends on its NaCl salt content to obtain its signature flavor and quality traits. This study evaluated the effects of different Na-K salt blends on the compositional, sensorial, microbial, functional, and rheological properties of QF stored for up to 12 wk at 4°C. Queso Fresco curd from each vat was divided into 6 portions and salted with different blends of NaCl-KCl (Na-K, %): 0.75-0.75, 1.0-0.5, 1.0-1.0, 1.0-1.3, 1.0-1.5, and 2.0-0 (control). Within this narrow salt range (1.5 to 2.5% total salt), the moisture, protein, fat, and lactose levels; water activity; pH; and the textural and rheological properties were not affected by salt treatment or aging. The total salt, sodium, potassium, and ash contents reflected the different Na-K ratios added to the QF. Total aerobic microbial count, overall proteolysis, the release of casein phosphopeptides, and the level of volatile compounds were affected by aging but not by the salt treatment. Only the 1.0-1.3 and 1.0-1.5 Na-K cheeses had sensory saltiness scores similar to that of the 2.0-0 Na-K control QF. Loss of free serum from the cheese matrix increased steadily over the 12 wk, with higher losses found in QF containing 1.5% total salt compared with the higher Na-K blends. In conclusion, KCl substitution is a viable means for reduction of sodium in QF resulting in only minor differences in the quality traits, and levels of 1.0-1.3 and 1.0-1.5 Na-K are recommended to match the saltiness intensity of the 2.0-0 Na-K control. The findings from this study will aid cheese producers in creating reduced-sodium QF for health-conscious consumers.

Key words: Hispanic-style cheese, reduced sodium

INTRODUCTION

Sodium chloride, NaCl, is an essential ingredient in cheese, affecting its flavor, functionality, texture, and safety (Guinee and Fox, 2004; Johnson et al., 2009). Yet, sodium in the diet has become a health issue over the past few years because excessive amounts may have negative associations with heart and kidney diseases, among others (USDA, 2010). Although debate continues as to how much sodium should be in the diet, many health-conscious consumers are interested in reducing their sodium intake for general wellness considerations, while others are required to reduce their sodium intake because of existing health concerns such as high blood pressure.

Although low-sodium foods may be viewed as healthier alternatives, a reduction in salt may have detrimental effects on the expected quality, including flavor and texture, of traditional products. Manufacture of Hispanic-style cheese in the United States has grown tremendously in the last 2 decades, representing 113 million kilograms or 2.2% of all cheese made in the United States in 2015 (NASS, 2016). Demand for the cheese has come from the growing Hispanic population and increased popularity of Latin-style foods in the United States. Queso Fresco (QF) is the variety most often sold in the United States, and it is characterized as a fresh, high-moisture cheese with relatively neutral pH (>6.0) and a NaCl content of 1 to 3% (Van Hekken and Farkye, 2003).

Reducing the sodium level in QF is challenging when simultaneously attempting to maintain the expected quality, including its signature salty flavor. Rapid chilling of QF after pressing and immediate vacuum packaging results in a commercial fresh cheese that maintains its quality for up to 8 wk. However, microbial and textural qualities of starter-free QF containing <1.0% NaCl were not maintained over an 8-wk storage period and were borderline acceptable at 1.0% NaCl (Guo et al., 2011, 2012). At 1% NaCl, QF contains about 30 to 50% less sodium than most commercial US-made QF that we have examined (unpublished data).
Food and Drug Administration regulations state that a 25% reduction in sodium is required for a cheese to be labeled as reduced-sodium, while cheese must not exceed 140 mg of sodium per serving size (50 g for cheese; about 0.7% NaCl) to be labeled as low-sodium (FDA, 2017). Without a federal standard of identity for QF, salt content is established by the manufacturer.

Consumers use saltiness as a defining taste attribute that influences their overall acceptance of a cheese (Schroeder et al., 1988), and they can detect a 25% reduction in salt in some cheeses (Johnson et al., 2009). Reducing NaCl also influences microbial growth and metabolism, which alters secondary proteolysis and can lead to bitterness and poor body development (Guinee and Fox, 2004). Salt substitutes, such as KCl, are typically added to cheese with lower sodium levels to improve salty notes, yet care must be taken to maintain the flavor profile. High concentrations of KCl have been shown to improve salty notes, yet care must be taken to maintain the flavor profile. High concentrations of KCl have been reported to introduce an acid or bitter off-flavor, while other salt substitutes add a savory note that may be acceptable in some aged cheeses but would not be appropriate in a fresh cheese such as QF. Grummer et al. (2012) reported that water activity ($a_w$) typically added to cheese with lower sodium levels to improve salty notes, yet care must be taken to maintain the flavor profile. High concentrations of KCl have been reported to introduce an acid or bitter off-flavor, while other salt substitutes add a savory note that may be acceptable in some aged cheeses but would not be appropriate in a fresh cheese such as QF. Grummer et al. (2012) reported that water activity ($a_w$) in reduced-sodium cheese using mineral replacers must be matched to full NaCl control to minimize differences in quality.

Because KCl is less effective than NaCl in inhibiting microbial growth (Guinee and Fox, 2004), the proteolysis and pH during matrix development may be altered and change many of the quality properties of the cheese. Compared to NaCl controls, the use of KCl in brine resulted in higher proteolysis in Akawi (Ayyash et al., 2012) but not in feta (Katsiari et al., 2000). Grummer et al. (2012) reported that water activity ($a_w$) in reduced-sodium cheese using mineral replacers must be matched to full NaCl control to minimize differences in quality.

To determine the effects of reducing the NaCl content to 0.75 or 1.0% and adding KCl to obtain total salt levels of 1.5 to 2.5% on the quality characteristics of starter-free QF over 12 wk of 4°C storage. A narrow 1.5 to 2.5% total salt range was selected to test saltiness at two 1:1 NaCl:KCl (Na-K) ratios, 0.75-0.75 and 1.0-1.0; reduce the NaCl level to 1.0% where quality traits began to change; and set at a maximum of 1.5% added KCl, a level at which off-flavors began to be detected. A 1.0-1.3 Na-K blend was included to match the $a_w$ of the 2.0-0 Na-K control.

**Materials and Methods**

**Cheesemaking**

Starter-free QF ($n = 3$) were made as described by Guo et al. (2011) with modifications to salt treatments. Briefly, milk (200 L) that had been homogenized, pasteurized, warmed to 32°C, and treated with CaCl$_2$ was coagulated with chymosin (13 mL/100 L of milk; Chymax, Chr. Hansen Inc., Milwaukee, WI); no starter culture was added. Curd was cut after 30 min, cooked at 39°C for 30 min, separated from the whey by draining the whey from the vat, chilled to 15°C, and fine milled. Curd was then divided into six 5-kg portions and received 1 of 6 salt treatments (wt/wt): NaCl (Morton Salt, Chicago, IL) was maintained at 1.0% and KCl (Morton Salt) was added at 0.5, 1.0, 1.3, and 1.5%; a 2% NaCl control and a 0.75% NaCl-0.75% KCl QF were also made (1.0-0.5, 1.0-1.0, 1.0-1.3, 1.0-1.5, 2.0-0, and 0.75-0.75 Na-K, respectively). All Na-K blends met the legal definition for reduced-sodium cheese. Curds for each treatment were packed into individual molds (10 cm × 20 cm × 25 cm) and horizontally pressed at 200 kPa for 12 h at 22°C. The next morning, as each treatment was removed from the mold, an interior portion of the cheese was collected aseptically for division and vacuum packing of samples for microbiology assays, another portion was removed for cutting and vacuum packing for the free serum study, and the remaining cheese was divided into 4 blocks to be stored at 4°C for up to 12 wk. At wk 1, 4, 8, and 12, 1 block from each salt treatment was removed from storage and divided for various assays.

**Chemical and Physiochemical Properties**

**Composition, pH, and Water Activity.** Cheese composition, water activity, and pH were measured at wk 1, 4, 8, and 12 as described by Guo et al. (2011). Moisture ($n = 3$) was determined using the forced-draft oven method (no. 948.12; AOAC International, 2000). Fat content ($n = 2$) was quantified based on a modified Babcock procedure (Kosikowski and Mistry, 1997). Total nitrogen levels ($n = 2$) were determined using a nitrogen analyzer (Flash EA1112; Thermo Fisher Scientific, Lakewood, NJ) and multiplied by 6.28 to obtain protein content. Lactose levels ($n = 3$) were determined using a lactose analyzer (Application Note 320, model YSI 2700 Select, YSI US, Yellow Springs, OH). Ash samples ($n = 3$) were generated using the forced-draft oven method (Lindberg Furnace, Watertown, WI) (no. 945.46; AOAC International, 2000). Initial salt levels ($n = 2$) were measured using high-range chloride titrator strips (Hach Co., Loveland, CO), and...
specific sodium and potassium levels were quantified on ash samples dissolved in 2% nitric acid (n = 2) using an inductively coupled plasma spectrometer (ICAP 6300 Duo, Thermo Fisher Scientific, Lakewood, NJ). Salt in moisture (S/M) was calculated using [total salt (from chloride titrator strips)/% moisture] × 100. An Aqualab CX2 water activity unit (Decagon Devices Inc., Pullman, WA) was used to record aw (n = 3) of the samples. A pH meter equipped with a point tip probe (model 611; Orion Research Corp., Cambridge, MA) was used to measure pH within the cheese block (n = 6).

**Proteolysis.** Protein profiles were created to track proteolysis of major caseins using SDS-PAGE protocols by Van Hekken et al. (2013a). Briefly, water-soluble proteins were isolated in the supernatant of casein samples homogenized (model 23, VirTis Co., Gardiner, NY) in a Tris-EDTA-SDS-dithiothreitol buffer and centrifuged at 39,000 × g for 60 min at 4°C. Lyophilized protein extracts were dissolved in a buffer containing Tris-EDTA-SDS-buffer (2× Laemmli sample buffer, BioRad, Hercules, CA) with added mercaptoethanol. Samples were separated on 20% homogeneous ultra-thin polyacrylamide gels (GE Healthcare Bio-Sciences Corp., Piscataway, NJ) using a PhastSystem (GE Healthcare Bio-Sciences Corp.). Protein bands were stained with Bio-Safe Coomassie (BioRad). Gels were scanned (V700 Photoscanner; Epson America, Inc., Long Beach, CA) and images analyzed using ImageQuant software (version 8.1; GE Healthcare Bio-Sciences Corp.).

**Casein Phosphopeptides.** Water-soluble protein samples extracted for PAGE were also used as the starting material to extract casein phosphopeptides (CPP) based on the methodology by Miquel et al. (2005) as modified by Tunick et al. (2016). Briefly, peptides were isolated in the supernatant of water-soluble protein samples that had been dissolved in Tris, pH 8, with CaCl2 and then ethanol added, and finally centrifuged at 1,200 × g for 30 min at 10°C. The resulting pellet was rinsed again with ethanol and centrifuged, and the 2 supernatants were combined with a small amount of water and lyophilized. Lyophilized peptides were dissolved in 0.1% formic acid and filtered to remove peptides larger than 10 kDa (Ultracel YM-10; EMD Millipore Corp., Darmstadt, Germany) before being injected into a reverse-phase ultra-performance liquid chromatography system (Nano-Acquity UPLC; Waters, Milford, MA) fitted with a trap column (Symmetry C18, 5 μm, 180 μm × 20 mm; Waters) and a separation column (Nano-Acquity HSS T3, 1.8 μm, 75 μm × 150 mm; Waters) and using a aqueous gradient of 5 to 40% acetonitrile/0.1% formic acid. A quadrupole time-of-flight analyzer mass spectrometer (Synapt G1; Waters) controlled by MassLynk Software (version 4.1; Waters) was used to further characterize separated peptides for identification using a milk proteins database from UniProt Knowledgebase (www.uniprot.org/help/uniprotkb), the ProteinLynk Global Server (version 4.1, Waters), and database search parameters of phosphorylation at serine sites (Ser-P), methionine oxidation, and nondigestion enzyme. Mass spectrometric data were analyzed in triplicate, and peptide sequences appearing in all 3 runs were determined to be real.

**Fatty Acid Profile.** Lipid profiles were created to track the shifts in fatty acid (FA) levels at wk 1, 4, 8, and 12 based on the procedure by Christie (2003). Briefly, liquid lipids collected from QF were converted to methyl esters by using hexane, sodium methoxide, glacial acetic acid, and CaCl2; after centrifugation, hexane was evaporated from the supernatant. Fatty acids were suspended in ethyl acetate and injected (n = 2) into a gas chromatograph (HP 6890; Hewlett-Packard, Santa Clara, CA) fitted with a SP-2380 fused silica capillary column (60 m × 0.25 mm; Supelco, Bellefonte, PA) and equipped with flame ionization detector. The reference standards included C4:0–C24:0 methyl esters and conjugated methyl linoleate (GLC448 and UC-59M, respectively; Nu-Chek Prep, Elysian, MN).

**Volatiles.** Volatile compounds were extracted from cheese and evaluated using GC-MS as described by Tunick et al. (2012). Briefly, finely chopped cheese (5 g) was sealed in a vial with 10 μL of 100-mg/kg 2-methyl-3-heptanone (Sigma Aldrich, St. Louis, MO) internal standard. The vial was warmed at 60°C for 10 min before a SPME fiber (50/30 μm DVB/Carboxen/PDMS Stableflex, Supelco) was inserted to absorb volatile compounds for 30 min. The fiber was then inserted into a GC-MS (7890A GC/5975 MS detector; Agilent Technologies, Wilmington, DE) fitted with a DB-5 column (30-m, 0.25-mm i.d., 0.25-μm film thickness; Agilent Technologies) to deabsorb for 5 min into a splitless injector held at 250°C and at a flow of 1 mL helium/min. NIST Standard Reference Database 1A NIST/EPA/NIH Mass Spectral Library (NIST05) in the instrument’s ChemStation software was used to analyze MS results and identify volatiles.

**Microbiology.**

Cheese samples (approximately 15 g, 1 cm × 1 cm × 8 cm) were individually vacuum packed in sterile 3-mil nylon polyethylene pouches (Prime Source Vacuum Products, San Jose, CA), and stored at 4°C. Two randomly selected bags were chosen at wk 0, 2, 4, 6, 8, 10,
and 12 for microbiological analysis. Ten-gram samples were homogenized in 90 mL of sterile 2% sodium citrate with a Stomacher 400C (Seward Ltd., London, UK) at 230 rpm for 2 min. Serial dilutions of homogenate (1:10) were prepared in sterile sodium citrate buffer and used to inoculate agar media. Total aerobic bacteria were enumerated on plate count agar (PCA; Oxoid, Basingstoke, UK) incubated at 30°C for 48 h. Psychrophilic bacteria were screened for on PCA incubated at 5°C for 12 d. Following incubation, agar media were analyzed for the number of colony-forming units and colony morphology. Colony-forming unit counts were performed on 2 different dilution plates, with numbers ranging between 10 and 200 cfu/plate; the limit of detection was 100 cfu/g of cheese.

Molecular identification of bacterial isolates from the starter-free QF was performed on a minimum of 2 colonies with the same morphology after overnight growth in brain heart infusion broth. One microliter of the overnight culture was used as template for PCR amplification of 16S rDNA using eubacterial oligonucleotide primers EubA and EubB (Cottrell and Kirchman, 2000). The PCR products were cleaned with the QIAquick PCR Purification Kit (Qiagen Inc., Valencia, CA), and nucleic acid sequencing was performed using an ABI Prism 3730 (Perkin-Elmer, Wellesley, MA) DNA analyzer with ABI Prism Big Dye terminator cycle sequencing reagent. Sequences were analyzed using Sequencher 54.2 (Gene Codes Corp., Ann Arbor, MI) and compared with sequences available in GenBank using the National Center for Biotechnology Information BLASTN search program (www.ncbi.nlm.nih.gov/BLAST/).

**Sensory**

Evaluation of saltiness for QF made with Na-K blends was conducted on samples from one vat of cheese. Evaluations were conducted on d 6 after manufacture after microbial evaluation by a licensed laboratory to ensure food safety. A discussion group, which consisted of 6 panelists, each with over 50 h of training and experience in flavor descriptive analysis of Hispanic-style cheese, received a refresher training session on salt intensity levels in solutions and cheese the day before evaluating samples. Samples (2.5 cm³) cut the day of testing were placed in 3-oz capped cups and refrigerated; samples were warmed to room temperature for 30 min before evaluation. Flavor profiles analysis of samples (2 per treatment) for salt intensity was conducted using a universal Spectrum 15-point intensity scale with salt intensity references at 2.5 and 5 (Meilgaard et al., 1999) and screened for the presence of potassium off-flavors. A roundtable discussion was then held to reach a consensus on the detection and description of the off-flavors. Sensory analysis was conducted to confirm that Na:P blends approached the saltiness of the control (2% NaCl, intensity of 2) yet were not overwhelmed by potassium bitterness; consumer acceptance or preference of the saltiness of QF was not an objective of this study.

**Free Serum**

Loss of free serum (wheying off) from the cheese matrix was measured using a modification to the protocol described by Van Hekken et al. (2012). Multiple cheese blocks (approximately 20 g, 2 cm × 3 cm × 1.5 cm) were cut on d 1, weighed, individually vacuum packed, and stored at 4°C; 3 randomly selected bags were removed at wk 0, 1, 2, 4, 6, 8, 10, and 12. Cheese blocks were carefully removed from the bag, blotted dry, and weighed. Loss of free serum from the cheese matrix during storage was measured as percent weight loss (wt/wt).

**Rheology and Texture**

**Rheology.** Assays were conducted at wk 1, 4, and 8 as described by Van Hekken et al. (2013b). Viscoelastic rheological properties were determined by small-amplitude oscillatory shear analyses using an AR-2000 rheometer (TA Instruments, New Castle, DE). Briefly, 3 disks of cheese (25.4 mm in diameter and 4 to 5 mm thick) were glued to parallel aluminum plates. Frequency sweeps were run from 1 to 100 rad/s at 0.8% strain. Values for elastic and viscous moduli (G’ and G”, respectively) obtained at 10 rad/s (1.6 Hz) are reported.

**Texture.** Textural properties were determined by texture profile analysis using a TA.XT2 texture analyzer (Texture Technologies Corp., Hamilton, MA) operating at a crosshead speed of 100 mm/min. Four cheese cylinders (13–14 mm diameter and 13–14 mm high) were compressed twice by 75% to obtain values for hardness, springiness, and cohesion.

**Statistics**

Data were analyzed using a randomized split plot design to determine the effects of cheese (vat), aging (weeks of storage), salt treatment, and week × treatment interaction (version 9.4; SAS Institute Inc., Cary, NC). A PROC MIXED model was used to conduct ANOVA using a Bonferroni LSD test to determine significant differences (P < 0.05).
RESULTS AND DISCUSSION

Composition and Physiochemical Properties

Composition, pH, and Water Activity. At wk 1, all QF contained 56.3 ± 0.6% moisture, 20.6 ± 0.3% protein, 20.5 ± 0.2% fat, and 3.1 ± 0.1% lactose; no significant differences were noted among salt treatments or between wk 1 and 12 ($P > 0.05$). As expected, characteristics based on salt content, such as the total salt, sodium, potassium, and ash concentrations and S/M values, reflected the different added amounts of NaCl and KCl (Table 1). Although the $a_w$ did not differ ($P > 0.05$) among individual salt treatments, overall differences in $a_w$ were observed at wk 1, 0.963 to 0.969, and wk 12, 0.974 to 0.979 ($P < 0.05$). The pH of the samples remained very stable, pH 6.30 ± 0.01 throughout the study. This stability is expected for a cheese made from pasteurized milk and without added LAB starter cultures. Although bacteria not destroyed during pasteurization will persist in QF (see Microbiology section for more details), pH > 6.0 indicated no excessive growth that would have affected cheese quality and shelf life.

Composition of QF fell within the wide range reported for QF (Van Hekken and Farkye, 2003). Lack of significant shifts in major components because of changing salt levels reflected the small salt range used (1.5–2.5% total salt) and typical vat-to-vat variations.

The effect of NaCl reduction on cheese composition can vary depending on the type of cheese and the level and method of NaCl reduction and NaCl substitution. Several studies have reported no significant compositional changes when using the 1:1 Na:K substitutions for Cheddar (Fitzgerald and Buckley, 1985) and feta (Katsiari et al., 2000), while other studies have reported shifts in composition when working with highly salted (4–6% NaCl) or brined cheeses (Johnson et al., 2009).

Legal definitions for low- and reduced-sodium cheese are clearly stated (FDA, 2017); however, without a standard of identity for QF, the normal NaCl content is determined by the manufacturer, which can range from 1 to 3% NaCl (Van Hekken and Farkye, 2003). Our full-salt control was set at 2.0% NaCl to obtain the salty flavor expected in QF. All of our blended Na-K QF conformed to the minimum 25% NaCl reduction to be considered reduced-sodium cheese. The lowest blend, 0.75-0.75 Na-K, approached but did not meet the required 2.8 mg of Na/g of cheese (about 0.7% NaCl) to be labeled as a low-sodium cheese.

Table 1. Effect of salt on selected characteristics of Queso Fresco made with different Na-K salt blends at wk 1 and 12

<table>
<thead>
<tr>
<th>Na-K treatments (%)</th>
<th>Total salt&lt;sup&gt;2&lt;/sup&gt; (%)</th>
<th>Na&lt;sup&gt;1&lt;/sup&gt; (mg/100 g of curd)</th>
<th>K&lt;sup&gt;1&lt;/sup&gt; (mg/100 g of curd)</th>
<th>Moisture (%)</th>
<th>Ash (%)</th>
<th>S/M&lt;sup&gt;3&lt;/sup&gt;</th>
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<td>111&lt;sup&gt;bc&lt;/sup&gt;</td>
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<sup>1</sup>Means with the same superscript within a column are not significantly different ($P < 0.05$).

<sup>2</sup>Salt blends (% NaCl-KCl): 2.0-0, 0.75-0.75, 1.0-0.5, 1.0-1.0, 1.0-1.3, and 1.0-1.5.

<sup>3</sup>Total salt = % NaCl + % KCl.

<sup>4</sup>$a_w$ = water activity.
Migrating ahead of β-CN was a series of peptide bands derived from the caseins, which were divided into 3 ranges using major whey proteins as division points; ranges included 18.5 to 23 kDa (between β-LG and β-CN), 14.5 to 18 (α-LA to β-LG), and 10 to 14 kDa (leading edge to α-LA). Peptides <10 kDa were allowed to migrate off the gel to improve the resolution of the native caseins. An increase (4.2 ± 0.8%) in the small peptides appeared in the 10 to 14 kDa range. Although a shift in bands occurred within the remaining ranges over time, the relative quantity within the ranges had not changed (P > 0.05). Queso Fresco contained 55% moisture, therefore, about 5 to 7% of the minor whey proteins (>66 kDa) remained in the cheese and decreased slightly to 3 to 5% by wk 12.

Without addition of starter cultures and their associated enzymes, proteolysis that occurred in the QF was driven by the coagulant enzyme, other indigenous milk enzymes that were not inactivated by pasteurization, and enzymes from bacteria that either survived pasteurization or were postcontaminants to the open cheese vat. The αS1- and β-CN are more readily hydrolyzed by chymosin than αS2-CN and, once native κ-CN is cleaved by chymosin in the coagulation step, κ-CN (f 1–105) remaining in the curd is fairly resistant to hydrolysis (Upadhyay et al., 2004). The αs1-CN of QF may not have not been optimal for chymosin activity (Upadhyay et al., 2004), the spoilage bacteria that survived pasteurization persisted (see Microbiology section) and contributed proteolytic enzymes. Although results are dependent on the specific cheese, proteolytic activity of starter cultures are typically reduced at >1.5% NaCl and >5.0 S/M (Guinee and Fox, 2004). Reducing NaCl, thereby increasing bacterial activity in the cheese, results in unacceptable flavor (bitter peptides) and poor quality in Cheddar-type cheese (Schroeder et al., 1988). Reduction of NaCl using a 1:1 Na-K blend did not alter the degree of proteolysis in 16-wk Cheddar (Fitzgerald and Buckley, 1985) and feta (Katsiari et al., 2000) cheese.

**Casein Phosphopeptides.** An additional health benefit of proteolysis within the cheese matrix is the release of small phosphorylated peptides from caseins that are believed to form complexes with calcium in the human digestive tract and help in passive adsorption of minerals through the intestinal wall (Miquel et al., 2005). The key to the mineral binding function of CPP is the phosphorylated serine (Ser-P) found in bovine caseins. Several peptides were extracted from QF that were ≤3 kDa (27 AA or less) and contained the Ser-P (Table 2). Peptides are identified by their AA sequences (Farrell et al., 2004). Bovine β-CN contains 5 Ser-P, located in the hydrophobic N-terminal end, and it released 74% of all of the CPP identified in this study. The highest number of peptides were

![Figure 1. Protein profiles for Queso Fresco made with different Na-K blends at wk 1 (lanes 2, 4, and 6) and wk 12 (lanes 3, 5, and 7), low molecular weight standard (lane 1), and bovine milk standard (lane 8). Salt blends (% NaCl-KCl): 2-0 (lanes 2 and 3), 0.75-0.75 (lanes 4 and 5), and 1.0-1.5 (lanes 6 and 7).](image-url)
found at wk 1 and 8 and the lowest at wk 12; the 1.0-1.5 Na-K blend consistently gave the lowest number of CPP. The β-CN CPP were categorized into 2 groups. Predominant peptides (85%) contained Ser-P at position 35 with β-Casein (f 33–37) was common to all and was cleaved at different points within the (f 26–56) sequence. Remaining peptides (15%) contained Ser-P at positions 15, 17, 18, or 19 with β-CN (f 10–15) common to all and were cleaved from the sequence (f 1–35); 30% of the sequences contained Ser-P triplet at (f 17–19); the numbers were highest at wk 1 and lowest at wk 12. Bovine αS1-CN contained 9 Ser-P and released 21% of the CPP. About 96% of CPP included Ser-P at position 115 with αS1-CN (f 113–119) common to all and were cleaved from sequence (f 104–122). Only 2 other sequences were found in QF; Ser-P at positions 41 and 46 [αS1-CN (f 36–47)] and at position 75 [(αS1-CN (f 69–81)]. Peptides began to appear in wk 4 samples and increased with time. Although bovine αS2-CN contains 11 Ser-P, it is noted for being resistant to proteolysis during cheese ripening (Upadhyay et al., 2004), and only 5% of the identified CPP were derived from this protein. All CPP came from the (f 137–152) sequence with αS2-CN (f 140–149) in common; CPP did not appear until wk 12 and only in 2.0-0, 1.0-1.3, and 1.0-1.5 Na-K QF. No CPP were identified as com-

Figure 2. Distribution of casein and peptide bands in Queso Fresco made with different Na-K salt blends and aged at 4°C at (a) wk 1 and (b) wk 12. Although treatment means within a protein or peptide group were not different at the same age (P > 0.05), some significant differences (P < 0.05) were observed between (a) wk 1 and (b) wk 12 as shown by different lowercase letters above the bars. Salt blends (% NaCl-KCl): 2.0-0, black; 0.75-0.75, white; 1.0-0.5, horizontal lines; 1.0-1.0, gray; 1.0-1.3, checked; and 1.0-1.5, grid lines.
ing from κ-CN. The only Ser-P in bovine κ-CN is at position 149 and is found in κ-CN (f 106–169) peptide, which remains in the whey; the bulk of it would have been lost when whey was drained from the vat.

Although CPP can bind with barium, cobalt, chromium, iron, magnesium, nickel, selenium, and zinc, and complexes of CPP-Na-Zn can inhibit formation of hydroxyapatite crystals (FitzGerald, 1998), the bulk of research has focused on the high affinity of CPP for calcium; the interaction between CPP and potassium has not been reported. Complexes that CPP form with calcium remain soluble within pH conditions of the intestine (Miquel et al., 2005) and facilitates passive adsorption of minerals in the human intestinal tract (Erba et al., 2001). Studies have reported mineral binding capabilities for the following CPP: αS1-CN (f 46–58), αS2-CN (f 59–79), αS2-CN (f 1–32), β-CN (f 55–64), β-CN (f 1–25), and β-CN (f 33–48) (Meisel and FitzGerald, 2003). Only a few Ser-P peptides identified in our study had the triple Ser-P double Glu sequence noted to have the highest affinity for binding minerals (Meisel and FitzGerald, 2003; Miquel et al., 2005). Further research is needed to determine mineral binding capacities for the Ser-P peptides isolated in this study and to evaluate their potential to enhance the bioavailability of different minerals.

### Fatty Acid Profile

Lipids make up 20% of QF and are essential to many quality traits, from matrix formation and stability to formation of flavor and aroma compounds to nutritional- and health-value contributions. As expected for cheese made from pasteurized milk without added starter cultures and their contributing lipases, the distribution of the FA in the full fat QF cheese was stable (P < 0.05) among salt treatments and between wk 1 and 12. The means and standard deviations (mg of FA/g of lipid) for common FA were: 6:0, 2.29 ± 0.11; 8:0, 0.87 ± 0.11; 10:0, 2.85 ± 0.17; 12:0, 3.38 ± 0.15; 14:0, 10.5 ± 0.5; 14:1, 0.89 ± 0.05; 15:0, 1.00 ± 0.08; 16:0, 27.93 ± 0.84; 16:1, 1.44 ± 0.08; 17:1, 0.53 ± 0.11; 18:0, 11.48 ± 0.48; 18:1, 26.88 ± 0.79; 18:1 trans, 4.41 ± 0.24; 18:2, 3.00 ± 0.26; and CLA, 1.22 ± 0.08. Salt treatments used in this study did not affect the concentrations of C18 lipids, 18:0, 18:1, 18:1 trans, or 18:2, thought to affect human health. Fatty acid values are similar to those published in the USDA Nutrient Database for fresh QF containing 1.9% NaCl in milligrams of FA per gram of fat bases (USDA, 2016). In this study, the lower NaCl treatments (0.75 and 1.0%) did not show excessive lipolysis indicative of excessive growth of spoilage bacteria that survived pasteurization (see Microbiology section for details). Guo et al. (2011) reported loss of QF quality at levels ≤

<table>
<thead>
<tr>
<th>Salt blend</th>
<th>β-CN</th>
<th>αS1-CN</th>
<th>αS2-CN</th>
</tr>
</thead>
<tbody>
<tr>
<td>2.0-0</td>
<td>13¹</td>
<td>ND</td>
<td>ND</td>
</tr>
<tr>
<td>0.75-0.75</td>
<td>6⁴²</td>
<td>6⁴²</td>
<td>ND</td>
</tr>
<tr>
<td>1.0-0.5</td>
<td>ND</td>
<td>6⁴²</td>
<td>14²</td>
</tr>
<tr>
<td>1.0-1.0</td>
<td>20¹</td>
<td>ND</td>
<td>15¹</td>
</tr>
<tr>
<td>1.0-1.3</td>
<td>14²</td>
<td>ND</td>
<td>ND</td>
</tr>
<tr>
<td>1.0-1.5</td>
<td>ND</td>
<td>ND</td>
<td>ND</td>
</tr>
</tbody>
</table>

¹Salt blends (%, NaCl-KCl): 2.0-0, 0.75-0.75, 1.0-0.5, 1.0-1.0, 1.0-1.3, and 1.0-1.5.
²The majority of CPP that originated from casein contained Ser-P at position 35 and came from within the β-CN [fragment (f) 26–56] region; (R)INKKIEK FOSEE QQTEDELDQKHFPFAQTQ(S); sequence common to all peptides is underlined.
³The second largest group of CPP that originated from casein contained Ser-P at position 115 and came from within the αS1-CN (f 104–122) region; (K)YKVQPLEIV PNSAEER LHS(M); sequence common to all peptides is underlined.
⁴All CPP that originated from αS2-CN contained Ser-P at position 143 and came from αS2-CN (f 37–52) region, (K)KTV DMSEVETK KTR(L); sequence common to all is underlined.
⁵One to 3 sequences contained Ser-P at position 15 and came from the β-CN (f 1-35) region, (A)RELEELNV PGEIVES LSSSEESITRINKKIEKFQ(S(E); sequence common to all is underlined.
⁶ND = no peptides containing Ser-P and <28 AA (about 3 kDa) were detected in sample.
⁷One sequence contained Ser-P at position 41 and 46 and came from the αS1-CN (f 36–47) region, (E)KVNELSKDIGSE(S).
⁸One sequence contained Ser-P at position 75 and came from the αS1-CN (f 69-81) region, (S)EEIVPNŚVEQKHI(Q).
1.0% NaCl but did not examine the FA profile to track lipolysis. Torres and Chandan (1981) reported low lipolysis in QF but did not state salt content. Salt levels in semihard and hard cheeses made with pasteurized milk and added cultures are typically used to regulate microbial growth and release of lipases to carefully control lipolysis and ensure development of desired quality traits during aging.

**Volatile compounds.** Ten aromatic volatile compounds common in cheese and at levels high enough to be detected were tracked from wk 1 to 12 (Table 3). Although they appeared in all cheese samples throughout the study, considerable variation existed among salt treatments, making it difficult to determine trends outside of saying that 2.0-0 Na-K control usually had the highest levels and 1.0-1.0 Na-K blend had the lowest levels. Only pentanal decreased slightly over time (2 to 1 mg/kg). The other 9 compounds increased in concentration as cheese aged. The most prevalent volatile compounds at 12 wk of aging were 2-nonanone and 2-heptanone at 62 and 42 mg/kg, respectively, with dramatic increases noted between wk 8 and 12. Octanoic acid and n-decanoic acid tripled to 14 mg/kg by wk 12, whereas heptanal and nonanal peaked around 8 mg/kg at wk 12. Hexanal and 2-pentanone increased to peak at 3 to 4 mg/kg by wk 12, while octanal doubled to 1 mg/kg at wk 12. Six other compounds (1-pentanal, 2-ethyl-1-hexanol, decanal, 2-undecanone, hexanoic acid, and 2H-pyran-2-one) were detected by SPME, but their appearance was random, not always appearing in all samples, and they are not included in Table 3.

As expected, volatile compounds that contribute to the signature flavor of QF were not affected by salt blends used in this study. At 12 wk of aging, the number of volatile compounds identified in QF were fewer than the 40-plus compounds reported in fresh cheese made with starter cultures (Garde et al., 2007) and the 100-plus compounds reported in aged cheeses (Frank et al., 2004). Short-chain aldehydes typically appear and then disappear early during aging of cheese as compounds convert to alcohols and acids, although aldehydes such as nonanal, which are generated by the lipolysis of longer-chain FA, would remain longer (Curioni and Bosset, 2002). Straight-chain aldehydes can enhance cheese flavor at low concentrations with green and fruity notes, yet they become unpleasant when found at levels above their thresholds (Curioni and Bosset, 2002). Ketones are common in cheese, typically increase during ripening, and bring fruity-floral notes to the cheese as well as green, blue cheese (2-heptanone), and hot milk and musty (2-nonanone) aromas. Short- and mid-chain FA are major contributors to cheese flavor, with octanoic and decanoic acids typically associated with goaty or rancid notes that become unpleasant at higher concentrations. Although 10 different volatile compounds were identified in QF, none of the sensory panelists identified any off-flavors in cheese samples, except for K-associated acid off-flavor in the 1.0-1.5 Na-K QF.

### Microbiology

Total aerobic counts for all QF samples did not differ significantly during 12 wk of 4°C storage. Counts were observed to decrease by 0.6 to 1.5 log cfu/g at wk 2 for QF made with ≥2% total salt (Figure 3), with total aerobic counts for these cheeses ranging between 3.6 ± 0.8 (2.0-0 Na-K) and 3.9 ± 0.8 (1.0-1.3 Na-K) log cfu/g; while counts for cheeses containing 1.5% total salt were 5.6 ± 0.7 (0.75-0.75 Na-K) and 5.5 ± 0.7 (1.0-0.5 Na-K) cfu/g at wk 2. Bacterial counts increased between wk 2 and 10 by 1 to 3 log cfu/g, with final bacterial load differing by ≤0.6 log cfu/g for all cheeses at wk 12. Total aerobic counts at wk 12 ranged between 5.8 ± 0.6 (2.0-0 Na-K) and 6.5 ± 1.1 (0.75-0.75 Na-K). Increased bacterial counts at 12 wk of storage was not unexpected because the cheeses in this study maintained S/M and aw values (Table 1) capable of supporting bacterial growth in cheese.

Because the QF was starter-free, molecular identification of bacterial species present between wk 8 and

<table>
<thead>
<tr>
<th>Compound</th>
<th>Lowest (mg/kg)</th>
<th>At wk</th>
<th>Highest (mg/kg)</th>
<th>At wk</th>
<th>Associated with odor or aroma¹</th>
</tr>
</thead>
<tbody>
<tr>
<td>2-Pentanone</td>
<td>0.1</td>
<td>1</td>
<td>4.0</td>
<td>12</td>
<td>Fruity, orange</td>
</tr>
<tr>
<td>2-Nonanone</td>
<td>0.1</td>
<td>1</td>
<td>62</td>
<td>12</td>
<td>Fruity, hot milk, musty</td>
</tr>
<tr>
<td>2-Heptanone</td>
<td>0.1</td>
<td>1</td>
<td>42</td>
<td>12</td>
<td>Green, fruity, blue cheese</td>
</tr>
<tr>
<td>Heptanal</td>
<td>0.5</td>
<td>1</td>
<td>8.0</td>
<td>12</td>
<td>Soapy, rancid</td>
</tr>
<tr>
<td>Octanal</td>
<td>0.5</td>
<td>1</td>
<td>1.0</td>
<td>12</td>
<td>Green, fatty</td>
</tr>
<tr>
<td>Pentanal</td>
<td>1.0</td>
<td>12</td>
<td>2.0</td>
<td>1</td>
<td>Chemical, green</td>
</tr>
<tr>
<td>Hexanal</td>
<td>2.0</td>
<td>1</td>
<td>4.0</td>
<td>8</td>
<td>Green, fruity</td>
</tr>
<tr>
<td>Nonanal</td>
<td>3.0</td>
<td>1</td>
<td>8.0</td>
<td>12</td>
<td>Green, tallow, soapy</td>
</tr>
<tr>
<td>Octanoic acid</td>
<td>4.0</td>
<td>1</td>
<td>14</td>
<td>12</td>
<td>Goat, musty, rancid</td>
</tr>
<tr>
<td>n-Decanoic acid</td>
<td>5.0</td>
<td>1</td>
<td>14</td>
<td>12</td>
<td>Goat, rancid</td>
</tr>
</tbody>
</table>

¹Curioni and Bosset, 2002; Le Quéré and Molimard, 2002.
12 was carried out by 16S rDNA sequencing, which identified *Paenibacillus* sp., *Pseudomonas putida*, *Microbacterium* sp., and *Bacillus simplex* in all cheese samples. These results suggest that variations in salt composition did not select for the growth of a specific contaminant. These bacterial species have all been reported as psychrotrophic contaminants capable of milk and dairy product spoilage (Hantsis-Zacharov and Halpern, 2007; Ranieri et al., 2012). Previously Guo et al. (2012) reported that QF made with ≤0.5% NaCl had significantly higher total aerobic counts compared with QF made with 1.0, 1.5, or 2.0% NaCl after 8 wk of storage. Similar counts observed for all cheeses in this study after 12 wk of storage suggested that replacement of NaCl with KCl did not result in a more favorable environment for bacterial growth. Suboptimal growth temperature and lack of oxygen due to vacuum packaging are expected to have negatively affected the growth of the aerobic psychrotrophs identified, thus contributing to the acceptable cheese quality maintained throughout the study. In addition, psychrophilic bacteria, which have a lower optimal growth temperature than psychrotrophs, were not observed on PCA incubated at 5°C. If present, they might have caused a more rapid degradation in cheese quality.

Another major concern with altering the type and concentration of salt used during the production of QF is the effect on food safety. High salt concentration used during QF manufacture is expected to prevent the growth of food-borne pathogens that would thrive in high moisture and neutral pH environment of QF, including *Listeria monocytogenes*. *Listeria monocytogenes* outbreaks have been associated with consumption of contaminated QF. A previous study reported that inoculated *Listeria* population increased by more than 4 log cfu/g in QF made with ~1.7% NaCl after 20 d of storage at either 4 or 10°C (Leggett et al., 2012). Another study reported that KCl could be used to replace NaCl without affecting the microbiological safety of food products (Boziaris et al., 2007). Taken together, these studies suggest that changes in salt composition tested in this study would not affect the growth or survival of *Listeria*. Because the main goal of this study was to demonstrate that the use of KCl would not affect QF quality, the cheese was not artificially contaminated with *Listeria*. Additional studies could be performed to test the survival of *Listeria* in Na-K blends of QF, but additional antilisterial treatments will most likely be required to suppress its growth.

**Sensory**

Based on earlier studies, a score of 2 for saltiness (based on the Spectrum intensity score of 2.5 for 0.2% NaCl in water; Meilgaard et al., 1999) was selected as the target saltiness for QF, which corresponded to the control cheese, 2.0-0 Na-K. Potassium chloride was selected to partially replace NaCl because of their chemical similarities, including the ability to provide a “salty” flavor note. Total replacement of NaCl with KCl is not recommended because it results in soft, bitter cheese with an unpleasant acid or bitter flavor (Fitzgerald and Buckley, 1985). Only one panelist detected a slight bitter potassium-associated off-flavor in the 1.0-1.5 Na-K QF. Of the Na-K blends tested, only QF containing 1.0-1.3 and 1.0-1.5 Na-K had saltiness intensities comparable to the control QF. Addition of 1.5% KCl should be the highest to be considered because potassium off-flavors would overwhelm the QF fresh milky flavor. Although a ratio of 1:1 Na:K has been suggested when using Na-K blends in cheese (Fitzgerald and Buckley, 1985; Lindsay et al., 1982), levels of 0.75-0.75 and 1.0-1.0% Na-K in QF were inadequate to mimic the saltiness of 2.0% NaCl. A study in the early 2000s reported that consumers on the west coast of the United States preferred QF that contained 1.4 to 2.4% salt, with consumers familiar with Latin-style cheeses preferring higher ranges; the optimum was suggested at 1.8% salt and pH 6.0 (Clark et al., 2001). Sensory evaluations definitely demonstrate that Na-K blends of 1.0-1.3 and 1.0-1.5 had the saltiness expected in QF without being overwhelmed by potassium bitterness, yet they should only be considered as guidance for future work because our sensory evaluation was based on limited testing. Further descriptive analysis of saltiness and off-flavors in fresh and aging QF should be conducted for full...
flavor profiles as well as tests of consumer difference, preference, and acceptance of reduced-sodium QF.

**Free Serum**

The amount of free serum (wheying off) lost from the cheese matrix and accumulated in the vacuum bag increased ($P < 0.05$) over time for all salt treatments (Figure 4). Queso Fresco containing 0.75-0.75 and 1.0–0.5 Na-K consistently had the highest loss of whey (8.2% at wk 12) of all salt treatments, whereas QF with 1.0–1.0 Na-K had the lowest loss of whey (6.8% at wk 12). Although 2.0-0 and 1.0-1.3 Na-K were matched for $a_w$, 1.0–1.3 Na-K QF was better at retaining whey within the cheese matrix than the 2.0-0 Na-K control. The amount of whey lost in this assay does not reflect what happens in larger cheese blocks. Smaller pieces used to determine wheying off and the fragile nature of the QF matrix elevated the percent loss over 12 wk of the study. Larger blocks of QF did have some whey accumulating in the bag during storage but only a few milliliters, which resulted in slight ($P > 0.05$) decreases in moisture content of 0.5 to 1.2% (Table 1).

Sodium chloride is an important factor in inducing syneresis in the curd and texture development (McMahon, 2010). Earlier studies have shown that wheying off is a problem for QF. Overnight pressing did not prevent wheying off the first week of storage (2–4% lost).

**Rheology and Texture**

**Rheology.** In small-amplitude oscillatory shear analyses, elastic modulus ($G'$) is a measure of the energy stored and recovered per oscillation, and viscous modulus ($G''$) is a measure of the energy dissipated and lost as heat per oscillation. The $G'$ and $G''$ values for all salt treatments were initially between 26.6 and 32.4 kPa, and $G''$ values were from 7.19 to 8.6 kPa (Figure 5). By 8 wk, $G'$ values had increased to 29.6 to 38.3 kPa and $G''$ values ranged from 7.44 to 9.14 kPa. Although values tended to increase with aging and 1.5% total salt QF were lower at wk 4 and 8 than other QF, differences were not significant ($P > 0.05$). The $G'$ and $G''$ values are relatively low for cheese, indicating that bonds between particles are being made and broken during each observation time, either spontaneously or from applied pressure.

![Figure 4](image-url) Distribution of accumulated free whey in packaging of Queso Fresco made with different Na-K salt blends over 12 wk of 4°C storage; standard error bars are included. Salt blends (% NaCl-KCl): 2.0-0: ●; 0.75-0.75: Δ; 1.0-0.5: ▲; 1.0-1.0: ◊; 1.0-1.3: ■; and 1.0-1.5: □.

![Figure 5](image-url) Means for viscoelastic properties [(a) elastic, $G'$ and (b) viscous, $G''$ moduli] of Queso Fresco made with different salt blends over 12 wk of 4°C storage; standard error bars are included. Salt blends (% NaCl-KCl): 2.0-0: ●; 0.75-0.75: Δ; 1.0-0.5: ▲; 1.0-1.0: ◊; 1.0-1.3: ■; and 1.0-1.5: □.
forces (Tunick, 2011). These results are expected for QF, which has a granular and crumbly nature. Over time, loss of whey caused casein strands in the matrix to become drier and less mobile yet had minimal effect on viscoelastic properties.

**Texture.** In texture profile analysis, hardness is the amount of force required to compress a specimen, springiness is the height of recovery after the first compression (or “bite”), and cohesiveness is the ratio of the force-time areas from the 2 compressions (Tunick, 2000). At wk 1, 1.0-1.5 Na-K QF was harder than 1.0-0.5 Na-K QF (31.8 and 25.9 N, respectively; *P* < 0.05), while no differences were noted among salt treatments for springiness, cohesiveness, and chewiness (7.07 ± 0.15 mm, 0.170 ± 0.005, and 34.7 ± 0.275, respectively) (Figure 6). Differences between salt treatments did not substantially change texture within the first 4 wk of storage. At wk 8, although no differences were noted among the salt treatments, hardness and chewiness had decreased to 22.0 ± 1.73 N and 34.7 ± 2.75 mJ, respectively (*P* < 0.05), and cohesiveness had increased to 0.186 ± 0.009 (*P* < 0.05). With storage, whey was forced out of the cheese with the first bite, which reduced force-time area, the force required to compress, and the recovery height. Hardness therefore decreased. Cohesiveness increased because force-time area of the second bite was relatively constant over time, and resulted in an increase in the ratio of the 2 force-time areas.

Sodium chloride, noted for enhancing protein hydration and solubility, especially κ-CN (f 1–105), contributes to forming tight and more continuous protein networks and affects curd formation (enhanced syneresis) and cheese rheology and texture (Guinee and Fox, 2004). Grummer et al. (2012) reported that *aw* of reduced-sodium cheese must be matched to full-salt cheese controls to minimize differences because of S/M. In that study, 2-mo-old Cheddar containing sea salt (22% Na, 9% K, 2% MgCl₂, and 23% sulfate) were the hardest, while 2.5% NaCl control and 1.2:1.7 Na:K were similar in hardness. The researchers concluded that the presence of the MgCl₂ in the sea salt was responsible for the firmer matrix. Total replacement of sodium with other minerals resulted in softer Cheddar cheese,

![Figure 6](image-url). Means for textural properties [(a) hardness, (b) springiness, (c) cohesiveness, and (d) chewiness] of Queso Fresco made with different Na-K salt blends over 12 wk of 4°C storage; standard error bars are included. Salt blends (% NaCl-KCl): 2.0-0, ●; 0.75-0.75, Δ; 1.0-0.5, ▲; 1.0-1.0, ○; 1.0-1.3, ■; and 1.0-1.5, □.
while partial replacement using salt blend affected protein and lipid breakdown without altering hardness (Fitzgerald and Buckley, 1985). In high moisture cottage cheese, reduction of NaCl by 50% did not alter consumer acceptability scores, but using KCl or Na-K blends to reduce the sodium to 50% lowered the quality of the cheese (Lindsay et al., 1982). When NaCl content was reduced in QF without using a salt replacer, the only significant trend was higher shear stress and shear rigidity of QF as NaCl levels increased (Guo et al., 2011). The strength of the matrix (stress required to fracture) in QF containing ≤ 1.0% NaCl decreased over time. Other than an increase at 8 wk in hardness as KCl concentration increased, the amount of Na-K did not affect rheology or texture of QF.

Results from this study show that lowering NaCl to 1.0% and varying KCl levels from 0.5 to 1.5% in QF did not alter major components (moisture, fat, and protein) of the cheese, did not cause major variation in the formation or stability of the cheese matrix, and resulted in a weak-bodied cheese that crumbled easily. Within the curd, no excessive microbial growth, proteolysis, or lipolysis occurred, which contributed to maintaining the quality traits QF during 12 wk of 4°C storage. Sensory evaluation targeted Na-K blends that matched 2.0% NaCl control saltiness intensity, although future research is needed to confirm levels and to test consumer acceptance of reduced-sodium QF using Na-K blends. Further research is also needed to assess the food safety issues that arise when lowering NaCl content to 1.0% in QF.

CONCLUSIONS

Reducing NaCl levels to 1% and adding 1 to 1.5% KCl had minimal effect on compositional, microbial, functional, and textural-rheological properties of QF, and adequate saltiness was obtained at 1.3 or 1.5% KCl. Reducing NaCl level to 1% and adding 1.3 or 1.5% KCl during cheesemaking showed that KCl substitution was a viable route for reducing sodium in QF, while retaining the desired quality traits and shelf life of the cheese. Results from this study will help in developing a reduced-sodium QF that meets the quality criteria of the cheese manufacturer and the health conscious consumer.

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